



7-Aryl-7-deazaadenine 2'-Deoxyribonucleoside Triphosphates (dNTPs): Better Substrates for DNA Polymerases than dATP in Competitive Incorporations**

Pavel Kielkowski, Jindřich Fanfrlík, and Michal Hocek*

Abstract: A series of 7-substituted 7-deazaadenine and 5-substituted cytosine 2'-deoxyribonucleoside triphosphates (dNTPs) were tested for their competitive incorporations (in the presence of dATP and dCTP) into DNA by several DNA polymerases by using analysis based on cleavage by restriction endonucleases. 7-Aryl-7-deazaadenine dNTPs were more efficient substrates than dATP because of their higher affinity for the active site of the enzyme, as proved by kinetic measurements and calculations.

Base-functionalized DNA has a wide range of applications and can be efficiently prepared by the incorporation of base-modified 2'-deoxyribonucleoside triphosphates (dNTPs) with polymerases, either by primer extension (PEX) or the polymerase chain reaction (PCR).^[1,2] These procedures use a modified dNTP in the absence of its natural counterpart to ensure that the polymerase incorporates the modified nucleotide. Little attention has been paid to incorporations of a modified nucleotide in competition with the natural one, presumably because it was supposed that the non-natural dNTPs must be much worse substrates for polymerases and result in a low frequency of incorporation. However, any prospective in vivo applications of modified dNTPs inherently require competitive incorporation in the presence of an excess of all four natural dNTPs. Therefore, we report here on the competitive incorporations of 5-modified dCTP and 7-modified 7-deaza-dATP derivatives (Figure 1).

Competitive in vivo incorporation of modified dNTPs is performed in DNA metabolic labeling, which traditionally uses 5-bromo-dUTP,^[3] and has recently been complemented by ethynyl- (EdU,^[4] EdC,^[5] or 7-ethynyl-7-deaza-dATP^[6]) or azido-substituted dNTPs,^[7] but no quantitative or kinetic study was reported. Detailed mechanistic and structural studies on the incorporation of modified dNTPs were

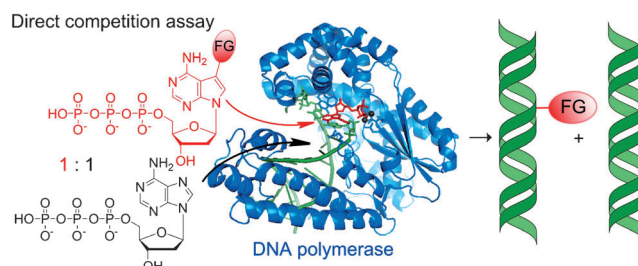


Figure 1. Competitive incorporation of dA^XTP versus dATP.

performed by Marx and co-workers,^[8,9] including a competitive study of single nucleotide incorporations, which were monitored by the different mobility of natural and modified extended primer in gel electrophoresis (PAGE). Most modified dNTPs were worse substrates for polymerases than the natural ones,^[8] with the exception of 7-[(hydroxydecanoyl)aminopentynyl]-7-deaza-dATP, which was a slightly better substrate for KlenTaq polymerase than dATP, but the authors did not comment on it further.^[9] However, this PAGE-based method can only be applied for the analysis of single nucleotide incorporation with bulky modifications, where the differences in mobility are distinctive. Therefore, a new method needed to be developed for the analysis of competitive primer extension and multiple incorporations (including PCR).

Recently, we studied the influence of major-groove modifications on the cleavage of DNA by restriction endonucleases (REs) and found that some REs do not tolerate any modifications, whereas other REs are blocked only by bulky groups.^[10] Therefore, we envisaged the use of REs^[11] for analyzing the products of competitive incorporations since, for any modification, we know specific REs which do not cleave the modified recognition sequence. Thus, PEX can be performed with a mixture of modified and natural dNTPs, followed by cleavage using the appropriate RE and PAGE analysis (Figure 2). This enables the proportion of modified to unmodified sequence to be quantified by the intensity of the two spots corresponding to the products.

We selected a series of eight previously reported 7-substituted 7-deaza-dNTPs and six 5-substituted dCTPs bearing various bulky functional groups. They included ethynyl and phenyl derivatives,^[10] redox labels (dN^{NO2}TP and dN^{NH2}TP),^[12] fluorescent labels (dA^{BFU}TP, dA^{ABOX}TP),^[13] reactive groups (dC^{FT}TP),^[14] and very bulky groups (dN^{STP}TP).^[15] They were incorporated through competitive PEX by different DNA polymerases (Pwo, KOD XL, Taq, Vent(*exo*-), Klenow, Bst, and human Pol α) to different

[*] P. Kielkowski, Dr. J. Fanfrlík, Prof. Dr. M. Hocek
Institute of Organic Chemistry and Biochemistry
Academy of Sciences of the Czech Republic
Gilead Sciences & IOCB Research Center
Flemingovo nám. 2, 16610 Prague 6 (Czech Republic)
E-mail: hocek@uochb.cas.cz
Homepage: <http://www.uochb.cas.cz/hocekgroup>

Prof. Dr. M. Hocek
Department of Organic Chemistry, Faculty of Science
Charles University in Prague
Hlavova 8, 12843 Prague 2 (Czech Republic)

[**] This work was supported by the ASCR (RVO: 61388963) and by the Czech Science Foundation (P206-12-G151).

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/anie.201404742>.

than those of natural dNTPs, thus showing a slower incorporation. However, the PEX (and even more so PCR) is a much more complex process, where the efficiency depends not only on the rate of incorporation but also on the presence or absence of proofreading (no correlation was observed for this factor) and on further extension of the primer.

To explain the higher affinity of the 7-aryl **dA^xTPs** to the active site of polymerases, we performed molecular modeling studies. Both dATP and **dA^{Ph}TP** were docked to the complex of Bst polymerase with primer and template using the known crystal structure (pdb 4BDP).^[18] The obtained tertiary complexes were studied using a semiempirical quantum mechanical scoring function.^[19] The resulting optimized structures are shown in Figure 3. The phenyl ring of **dA^{Ph}TP** filled the space

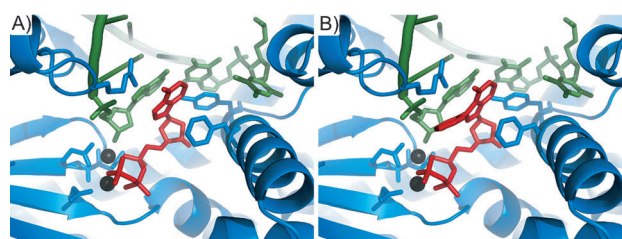


Figure 3. Bst polymerase active site with modeled A) dATP and B) **dA^{Ph}TP**.

between the aliphatic chain of Arg629 and the phenyl group of Phe710. The **dA^{Ph}TP** had a larger affinity (more negative score) than dATP (−72.8 and −65.0 kcal mol^{−1}, respectively), which indicates a possible increase in π - π stacking as a result of the 7-phenyl group. A similar binding mode and increased stacking is expected for other 7-aryl-7-deaza-dATP analogues and it is in accord with the high affinity of these **dA^xTPs** to this polymerase.

In conclusion, we have developed a method^[11] that is based on cleavage of the DNA by RE and PAGE analysis for the analysis of products generated from competitive incorporations of a modified **dN^xTP** in the presence of the corresponding natural dNTP. Our systematic study of competitive PEX using different **dA^xTPs** and **dC^xTPs** in different ratios with the natural dATP and dCTP revealed a surprising and counterintuitive result that most 7-aryl **dA^xTPs** are better substrates for most polymerases than dATP, whereas most **dC^xTPs** are worse substrates. The kinetic study and semiempirical calculations explain this by increased affinity of the 7-aryl **dA^xTPs** to the active site of the polymerase complex with the primer and template because of increased stacking. These findings are important for better understanding the mechanism of the incorporation of modified dNTPs by polymerase, as well as for the design of modified nucleosides and nucleotides for in vivo synthesis of base-modified DNA. Unfortunately, except for the known **dA^E** nucleoside,^[6] the 7-aryl **dA^x** nucleosides cannot be directly used for metabolic labeling, presumably because of poor intracellular phosphorylation to dNTPs and/or limited cellular uptake. The solution can be either synthesis of masked **dN^xTP** prodrugs or the

development of transport systems for **dN^xTPs** through cell membranes. These studies are currently under way.

Received: April 28, 2014

Published online: May 30, 2014

Keywords: DNA · enzymes · kinetics · nucleoside triphosphates · nucleotides

- [1] Reviews: a) M. Hocek, M. Fojta, *Chem. Soc. Rev.* **2011**, 40, 5802–5814; b) M. Hollenstein, *Molecules* **2012**, 17, 13569–13591.
- [2] Examples: a) S. Jäger, G. Rasched, H. Kornreich-Leshem, M. Engeser, O. Thum, M. Famulok, *J. Am. Chem. Soc.* **2005**, 127, 15071–15082; b) S. Obeid, M. Yulikov, G. Jeschke, A. Marx, *Angew. Chem.* **2008**, 120, 6886–6890; *Angew. Chem. Int. Ed.* **2008**, 47, 6782–6785; c) C. T. Wirges, J. Timper, M. Fischler, A. S. Sologubenko, J. Mayer, U. Simon, T. Carell, *Angew. Chem.* **2009**, 121, 225–229; *Angew. Chem. Int. Ed.* **2009**, 48, 219–223; d) N. Ramsay, A.-S. Jemth, A. Brown, N. Crampton, P. Dear, P. Holliger, *J. Am. Chem. Soc.* **2010**, 132, 5096–5104; e) J. Dadová, P. Orság, R. Pohl, M. Brázdová, M. Fojta, M. Hocek, *Angew. Chem.* **2013**, 125, 10709–10712; *Angew. Chem. Int. Ed.* **2013**, 52, 10515–10518; f) J. Balintová, M. Plucnara, P. Vidláková, R. Pohl, L. Havran, M. Fojta, M. Hocek, *Chem. Eur. J.* **2013**, 19, 12720–12731.
- [3] a) M. J. Bessman, I. R. Lehman, J. Adler, S. B. Zimmerman, E. S. Simms, A. Kornberg, *Proc. Natl. Acad. Sci. USA* **1958**, 44, 633–641; b) S. Fakan, *J. Microsc.* **1976**, 106, 159–171.
- [4] A. Salic, T. J. Mitchison, *Proc. Natl. Acad. Sci. USA* **2008**, 105, 2415–2420.
- [5] a) D. Ou, G. Wang, Z. Wang, L. Zhou, W. Chi, S. Cong, X. Ren, P. Liang, B. Zhang, *Anal. Biochem.* **2011**, 417, 112–121; b) L. Guan, G. van der Heijden, A. Bortvin, M. M. Greenberg, *ChemBioChem* **2011**, 12, 2184–2190.
- [6] A. B. Neef, F. Samain, N. W. Luedtke, *ChemBioChem* **2012**, 13, 1750–1753.
- [7] A. B. Neef, N. W. Luedtke, *ChemBioChem* **2014**, 15, 789–793.
- [8] a) S. Obeid, A. Baccaro, W. Welte, K. Diederichs, A. Marx, *Proc. Natl. Acad. Sci. USA* **2010**, 107, 21327–21331; b) K. Betz, F. Streckenbach, A. Schnur, T. Exner, W. Welte, K. Diederichs, A. Marx, *Angew. Chem.* **2010**, 122, 5308–5311; *Angew. Chem. Int. Ed.* **2010**, 49, 5181–5184; c) A. Baccaro, A.-L. Steck, A. Marx, *Angew. Chem.* **2012**, 124, 260–263; *Angew. Chem. Int. Ed.* **2012**, 51, 254–257; d) S. Obeid, H. Busskamp, W. Welte, K. Diederichs, A. Marx, *Chem. Commun.* **2012**, 48, 8320–8322; e) S. Obeid, H. Bußkamp, W. Welte, K. Diederichs, A. Marx, *J. Am. Chem. Soc.* **2013**, 135, 15667–15669; f) K. Bergen, K. Betz, W. Welte, K. Diederichs, A. Marx, *ChemBioChem* **2013**, 14, 1058–1062.
- [9] K. Bergen, A.-L. Steck, S. Strütt, A. Baccaro, W. Welte, K. Diederichs, A. Marx, *J. Am. Chem. Soc.* **2012**, 134, 11840–11843.
- [10] a) H. Macíčková-Cahová, M. Hocek, *Nucleic Acids Res.* **2009**, 37, 7612–7622; b) H. Macíčková-Cahová, R. Pohl, M. Hocek, *ChemBioChem* **2011**, 12, 431–438; c) P. Kielkowski, H. Macíčková-Cahová, R. Pohl, M. Hocek, *Angew. Chem.* **2011**, 123, 8886–8889; *Angew. Chem. Int. Ed.* **2011**, 50, 8727–8730; d) P. Kielkowski, N. Brock, J. Dickschat, M. Hocek, *ChemBioChem* **2013**, 14, 801–804.
- [11] Previously, REs were analogously used in the analysis of the fidelity of replication with artificial base pairs: a) F. Chen, Z. Yang, M. Yan, J. B. Alvarado, G. Wang, S. A. Benner, *Nucleic Acids Res.* **2011**, 39, 3949–3961; b) Z. Yang, F. Chen, J. B. Alvarado, S. A. Benner, *J. Am. Chem. Soc.* **2011**, 133, 15105–15112.

- [12] H. Cahová, L. Havran, P. Brázdilová, H. Pivoňková, R. Pohl, M. Fojta, M. Hocek, *Angew. Chem.* **2008**, *120*, 2089–2092; *Angew. Chem. Int. Ed.* **2008**, *47*, 2059–2062.
- [13] J. Riedl, R. Pohl, L. Rulišek, M. Hocek, *J. Org. Chem.* **2012**, *77*, 1026–1044.
- [14] V. Raindlová, R. Pohl, M. Šanda, M. Hocek, *Angew. Chem.* **2010**, *122*, 1082–1084; *Angew. Chem. Int. Ed.* **2010**, *49*, 1064–1066.
- [15] H. Macíčková-Cahová, R. Pohl, P. Horáková, L. Havran, J. Špaček, M. Fojta, M. Hocek, *Chem. Eur. J.* **2011**, *17*, 5833–5841.
- [16] a) A. H. Polesky, T. A. Steitz, N. D. F. Grindley, C. M. Joyce, *J. Biol. Chem.* **1990**, *265*, 14579–14591; b) H. Kong, R. B. Kucera, W. E. Jack, *J. Biol. Chem.* **1993**, *268*, 1965–1975; c) J. N. Patro, M. Urban, R. D. Kuchta, *Biochemistry* **2009**, *48*, 8271–8278.
- [17] J. G. Bertram, K. Oertell, J. Petruska, M. F. Goodman, *Biochemistry* **2010**, *49*, 20–28.
- [18] J. R. Kiefer, C. Mao, J. C. Braman, L. S. Beese, *Nature* **1998**, *391*, 304–307.
- [19] a) J. Fanfrlik, A. K. Bronowska, J. Rezac, O. Prenosil, J. Konvalinka, P. Hobza, *J. Phys. Chem. B* **2010**, *114*, 12666–12678; b) M. Lepsik, J. Rezac, M. Kolař, A. Pecina, P. Hobza, J. Fanfrlik, *ChemPlusChem* **2013**, *78*, 921–931.